

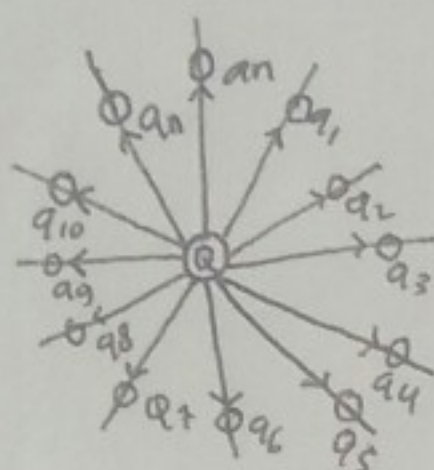
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 Kelas: B

$$\begin{aligned}
 1. N &= (5 - 0,8) \cdot \frac{6,02 \times 10^{23}}{107,9} + 0,8 \cdot \frac{6,02 \times 10^{23}}{197} \\
 &= 4,2 \cdot \frac{6,02 \times 10^{23}}{107,9} + 0,8 \cdot \frac{6,02 \times 10^{23}}{197} \\
 &= 2,34 \times 10^{22} + 2,44 \times 10^{21} \\
 &= 2,58 \times 10^{22}
 \end{aligned}$$

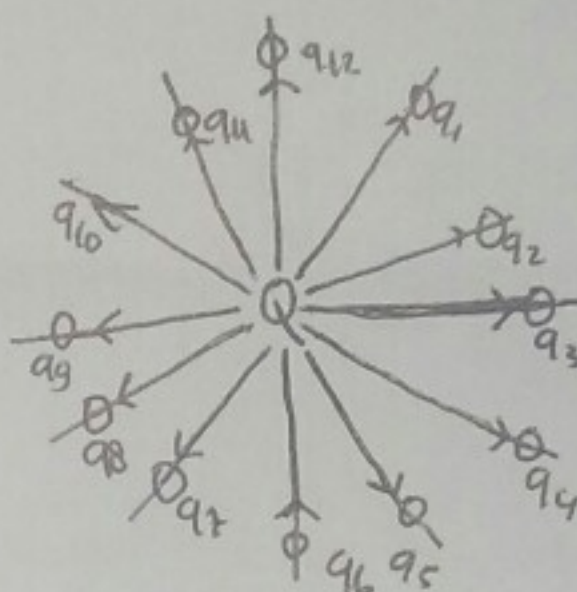
→ Atom Ag mempunyai 47 elektron atom Au mempunyai 79 elektron
 maka jumlah kedua atom tersebut adalah $47 + 79 = 126$ elektron,
 jadi jumlah medan total Q yaitu: $Q (2,58 \times 10^{22} \text{ atom})(126 \text{ elektron/atom})$
 $(1,60 \times 10^{-19} \text{ C/elektron}) = 5,2 \times 10^5 \text{ C}$, jadi untuk massacincin
 5 gr mempunyai muatan $5,2 \times 10^5 \text{ C}$, hal ini berarti dalam massa cincin
 5gr terdapat $\frac{5,2 \times 10^5}{1,6 \times 10^{-19}} = 3,25 \times 10^{24}$ elektron

* Kasus Gaya listrik

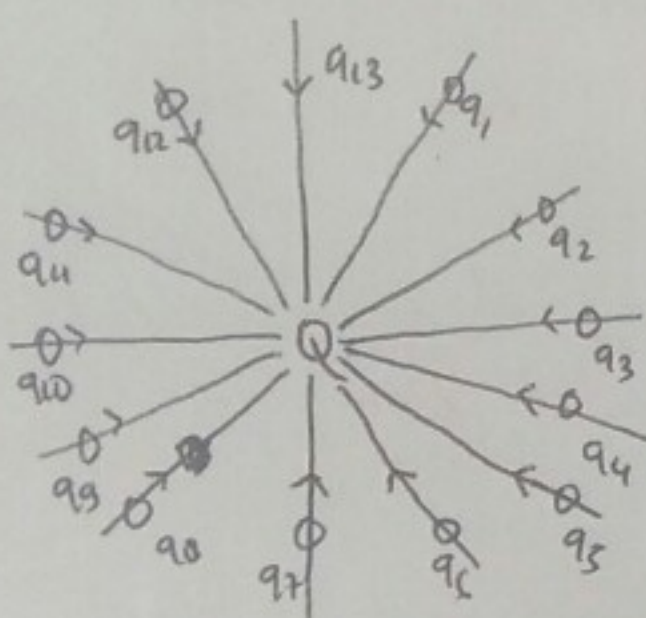
a.



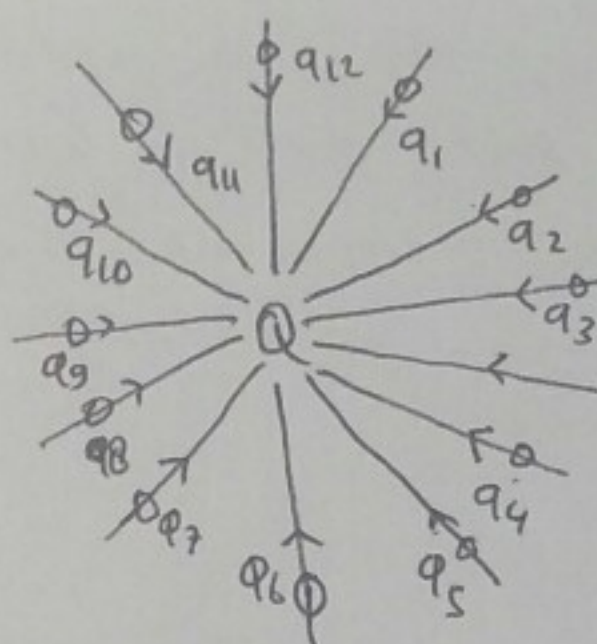
b.



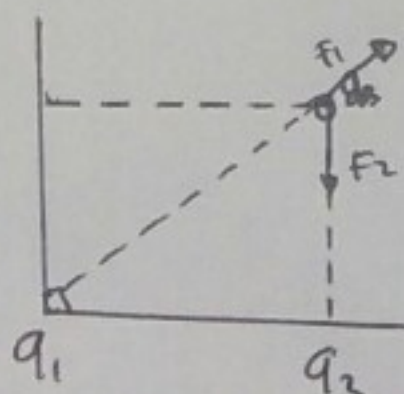
c.



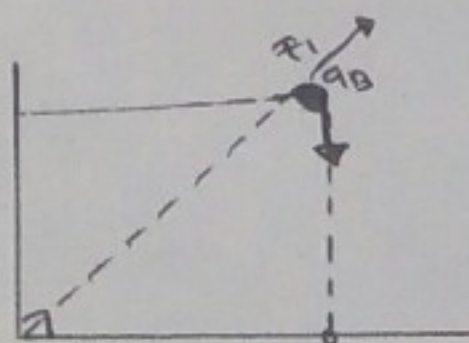
d.



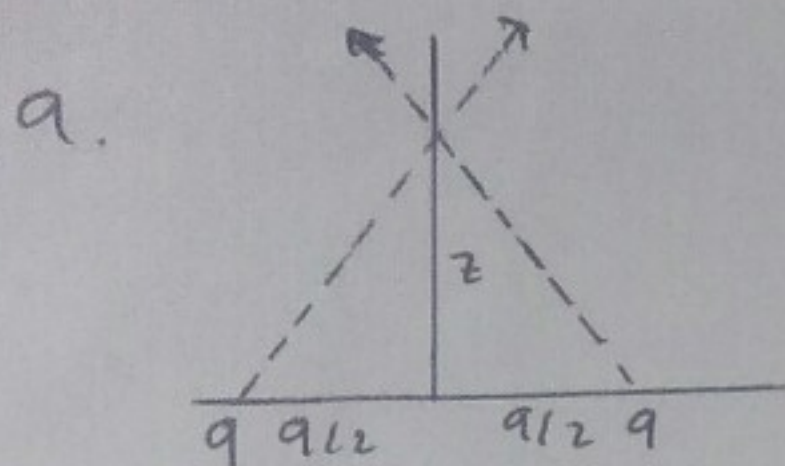
e. $q_1 = 25 \mu\text{C} \rightarrow (0,0)\text{m}$
 $q_2 = -10 \mu\text{C} \rightarrow (4,0)\text{m}$
 $q_3 = 10 \mu\text{C} \rightarrow (4,3)\text{m}$



f. $q_1 = 25 \mu\text{C} \rightarrow (0,0)$
 $q_2 = -15 \mu\text{C} \rightarrow (4,0)$
 $q_3 = 20 \mu\text{C} \rightarrow (4,3)$

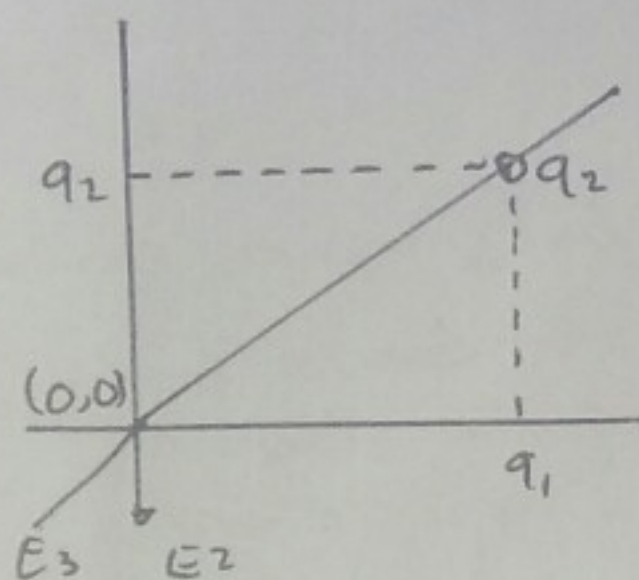


* Kasus Medan Listrik



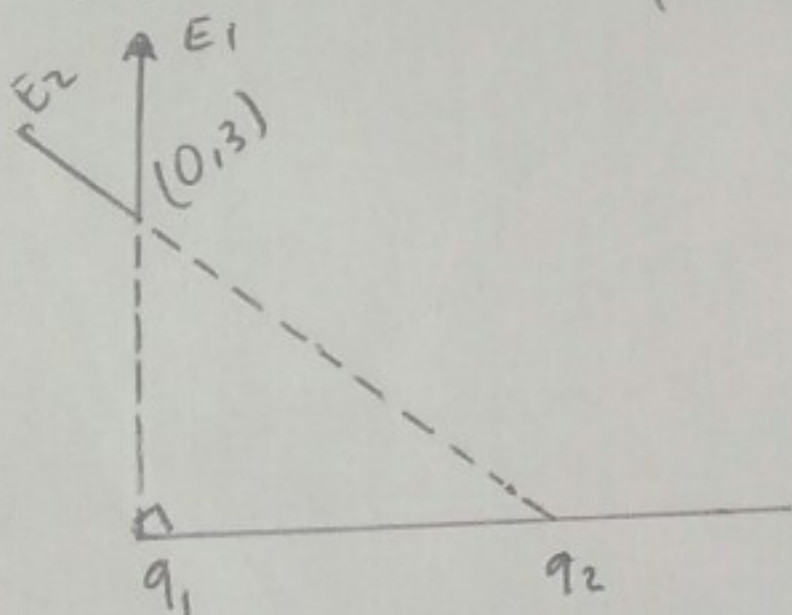
b. $q_1 (4,0) = 2 \text{ C}$
 $q_2 (0,3) = 5 \text{ C}$
 $q_3 (4,3) = 10 \text{ C}$

Medan listrik pada posisi $(0,0)$



c. $q_1 (0,0) = -5 \text{ uC}$
 $q_2 (4,0) = 10 \text{ uC}$

Medan listrik pada posisi $(0,3)$

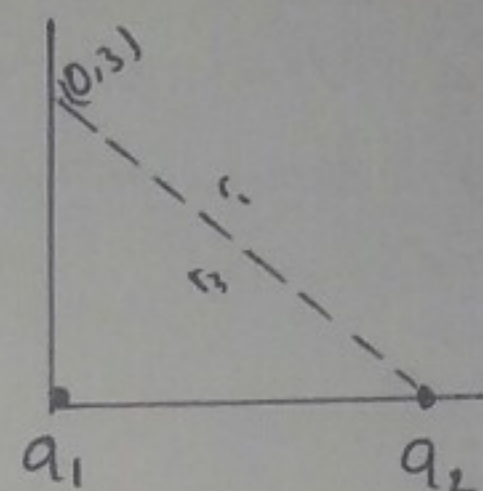


* Potensial listrik

2. $q_1 (0,0) = -5 \text{ C}$

$q_2 (4,0) = 10 \text{ uC}$

Hitung medan listrik titik $(0,3)$



$$V = \frac{kq_1}{r_1} + \frac{kq_2}{r_2}$$

$$= 9 \times 10^6 \left(\frac{5}{3} + \frac{10}{4} \right)$$

$$= \underline{\underline{3,75 \times 10^7}}$$